

INFRARED SPECTRA OF ANIONIC COBALT-CARBON DIOXIDE CLUSTERS

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We present infrared photodissociation spectra of $\text{Co}(\text{CO}_2)_n^-$ ions ($n = 3 - 11$) in the wavenumber region $1000 - 2400 \text{ cm}^{-1}$, interpreted with the aid of density functional theory calculations. The spectra show signatures of several structural motifs for the interaction of a Co atom and CO_2 ligands. The spectra are dominated by a core ion showing bidentate interaction of two CO_2 ligands forming C-Co and O-Co bonds. The prevalence of triplet vs singlet states and the charge distribution in the $\text{Co}(\text{CO}_2)_2^-$ core ion will also be discussed.